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Key indicators

Single-crystal X-ray study T = 294 KMean $\sigma(\text{C}-\text{C}) = 0.003 \text{ Å}$ R factor = 0.040 wR factor = 0.110 Data-to-parameter ratio = 13.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

(E)-1-(3-Ethoxy-4-hydroxybenzylidene)-2-(4-nitrophenyl)hydrazine

The molecule of the title compound, $C_{15}H_{15}N_3O_4$, is nonplanar. The ethylvanillin group makes a dihedral angle of 7.68 (7)° with the nitrophenylhydrazine mean plane. A trifurcated intra/intermolecular $O-H\cdots(O,O,O)$ hydrogenbond system helps to establish the molecular conformation and consolidate the crystal packing. Received 10 October 2005 Accepted 11 October 2005 Online 19 October 2005

Comment

One of the aims of investigating the structural chemistry of Schiff bases is to develop protein and enzyme mimics (Santos *et al.*, 2001). As part of an investigation of the coordination properties of Schiff bases functioning as ligands, we report the synthesis and structure of the title compound, (I) (Fig. 1).



The ethylvanillin group (C1–C9/O1/O2) in (I) is planar, with an r.m.s. deviation, δ , from the mean plane of 0.044 Å. The nitrophenylhydrazine moiety (C10–C15/N1/N2/N3) is also planar, with $\delta = 0.030$ Å. The dihedral angle between the two mean planes is 7.68 (7)°. The bond lengths and angles for (I) are unexceptional.

A trifurcated $O-H\cdots(O,O,O)$ intra/intermolecular $O-H\cdots O$ hydrogen-bond system is found in (I) (Table 1). The intramolecular bond stabilizes the conformation of the molecule, while the intermolecular bonds to the two O atoms of a nearby nitro group help to consolidate the crystal packing (Fig. 2). Conversely, the NH grouping does not participate in hydrogen bonds.





Figure 1 The structure of (I), with displacement ellipsoids for non-H atoms drawn at the 30% probability level.

Experimental

An anhydrous ethanol solution of 3-ethoxy-4-hydroxybenzaldehyde (1.66 g, 10 mmol) was added to an anhydrous ethanol solution of 1-(4-nitrophenyl)hydrazine (1.53 g, 10 mmol) and the mixture was stirred at 350 K for 5 h under nitrogen. A yellow precipitate appeared, which was isolated, recrystallized from ethanol, and then dried in a vacuum to give the pure compound in 87% yield. Yellow single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of an ethanol solution.

Crystal data

$C_{15}H_{15}N_{3}O_{4}$	$D_x = 1.398 \text{ Mg m}^{-3}$
$M_r = 301.30$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 2008
a = 10.148 (2) Å	reflections
b = 9.0980 (19) Å	$\theta = 2.6-25.9^{\circ}$
c = 15.967 (3) Å	$\mu = 0.10 \text{ mm}^{-1}$
$\beta = 103.868 \ (3)^{\circ}$	T = 294 (2) K
V = 1431.2 (5) Å ³	Block, yellow
Z = 4	$0.34 \times 0.30 \times 0.24$ mm
Data collection	

2904 independent reflections

 $R_{\rm int} = 0.040$

 $\theta_{\rm max} = 26.4^{\circ}$ $h = -12 \rightarrow 12$

 $k = -9 \rightarrow 11$

 $l = -19 \rightarrow 17$

1758 reflections with $I > 2\sigma(I)$

Bruker SMART CCD area-detector diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{\min} = 0.960, T_{\max} = 0.975$ 7792 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0459P)^2$
$R[F^2 > 2\sigma(F^2)] = 0.040$	+ 0.2257P]
$wR(F^2) = 0.110$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.00	$(\Delta/\sigma)_{\rm max} = 0.001$
2904 reflections	$\Delta \rho_{\rm max} = 0.18 \text{ e} \text{ Å}^{-3}$
209 parameters	$\Delta \rho_{\rm min} = -0.17 \text{ e } \text{\AA}^{-3}$
H atoms treated by a mixture of	Extinction correction: SHELXL97
independent and constrained	Extinction coefficient: 0.0170 (18)
refinement	

Table 1

Hydrogen-bond geometry (Å, $^{\circ}$).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$\begin{array}{c} O2 - H2 \cdots O1 \\ O2 - H2 \cdots O4^{i} \\ O2 - H2 \cdots O3^{i} \end{array}$	0.83 (3)	2.27 (2)	2.6840 (19)	111 (2)
	0.83 (3)	2.26 (3)	2.987 (2)	146 (2)
	0.83 (3)	2.49 (3)	3.205 (2)	145 (2)

Symmetry code: (i) $x - 1, -y + \frac{1}{2}, z + \frac{1}{2}$.

Figure 2 Hydrogen-bonding interactions (dashed lines) in (I).

The H atoms attached to O and N atoms were found in a difference map and their positions and $U_{\rm iso}$ values were freely refined. Other H atoms were included in calculated positions (C-H = 0.93-0.97 Å) and refined as riding with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$ or $U_{\rm iso}({\rm H}) = 1.5U_{\rm eq}({\rm methyl}\ {\rm C})$.

Data collection: *SMART* (Bruker, 1999); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1999); software used to prepare material for publication: *SHELXTL*.

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